



University of California, Los Angeles

Experimental study of flibe vapor condensation

Presented by:

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Town Meeting on IFE Liquid Wall Chamber Dynamics

Livermore, CA

May 5-6, 2003

Staged implementation of Flibe use

Flibe handling

Stage I Source testing

Transparent plastic chamber for easy visualization (low T)

Materials:

- plastic polycarbonate (Lexan)
- Teflon (CF_2 molecular chain)
- Metals (Tin, Al)
- Flibe



Stage II Diagnostic

SS chamber with view ports, insertion ports and heater for high T

Spectroscopy:

- Teflon (CF_2 molecular chain)
- LiF

Pressure history - non condensable gases:

- LiF



Stage III

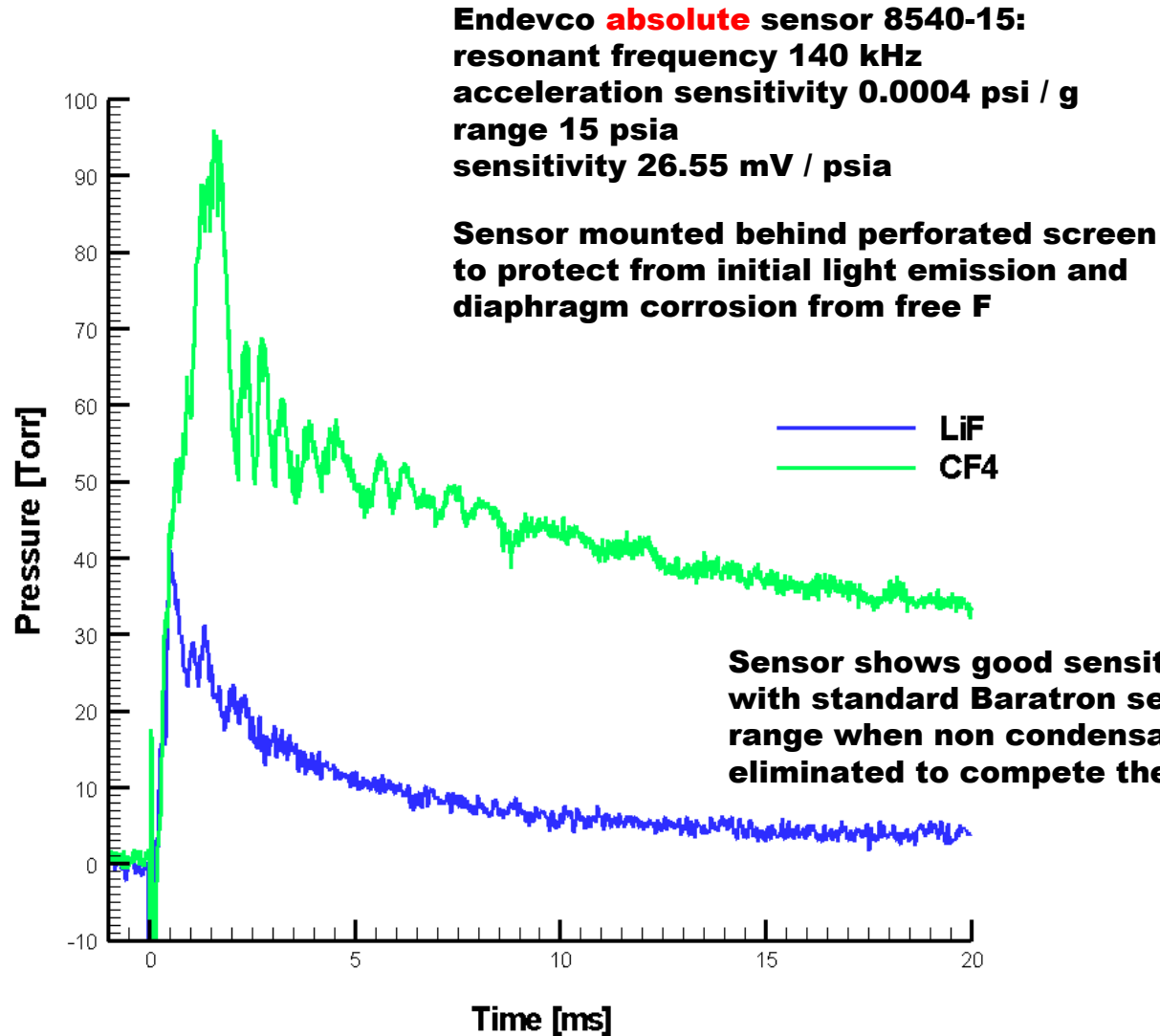
flibe

**$2/3 \text{LiF} + 1/3 \text{BeF}_2$
in moles**

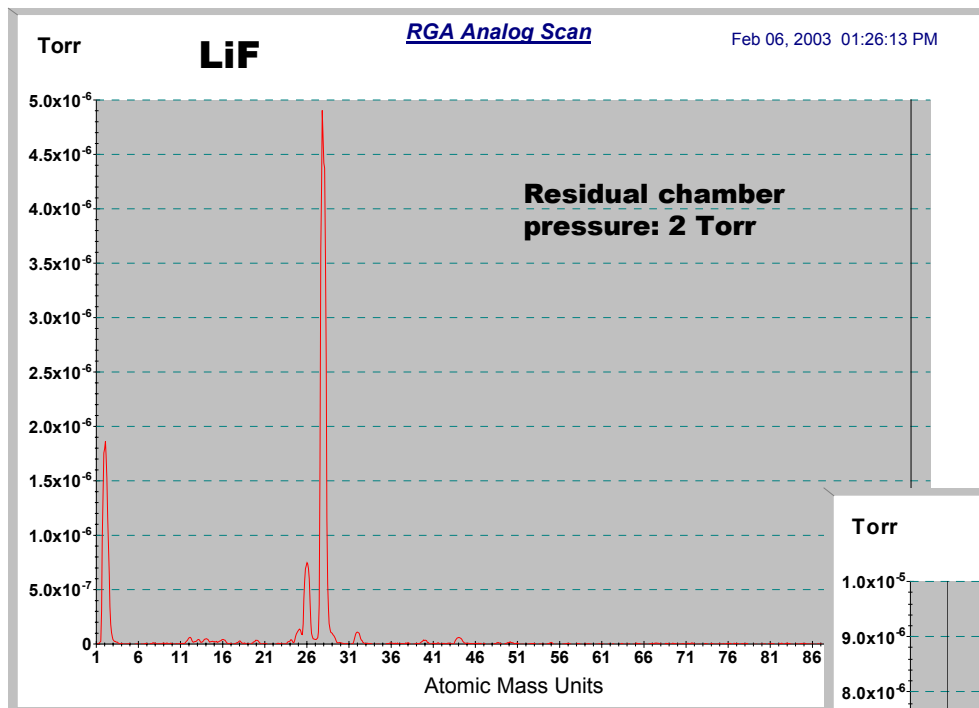
Total amount of purified flibe available for experiments:
1 liter from INEEL

Casting disks with $1/4$ inch central bore for plasma source sleeve

Chamber pressure history



Mass spectrometry for residual gases analysis



28 peak accounts in both cases for more than 50% of the total pressure. It is a complex peak that accumulates contributions from N_2 , CO , CO_2 and traces of more complex hydrocarbons such as ethane

Air (Nitrogen) can be evaluated by the 14 peak of dissociated N

Residual gas composition in the chamber for LiF shot evaluated with RGA correction factors:

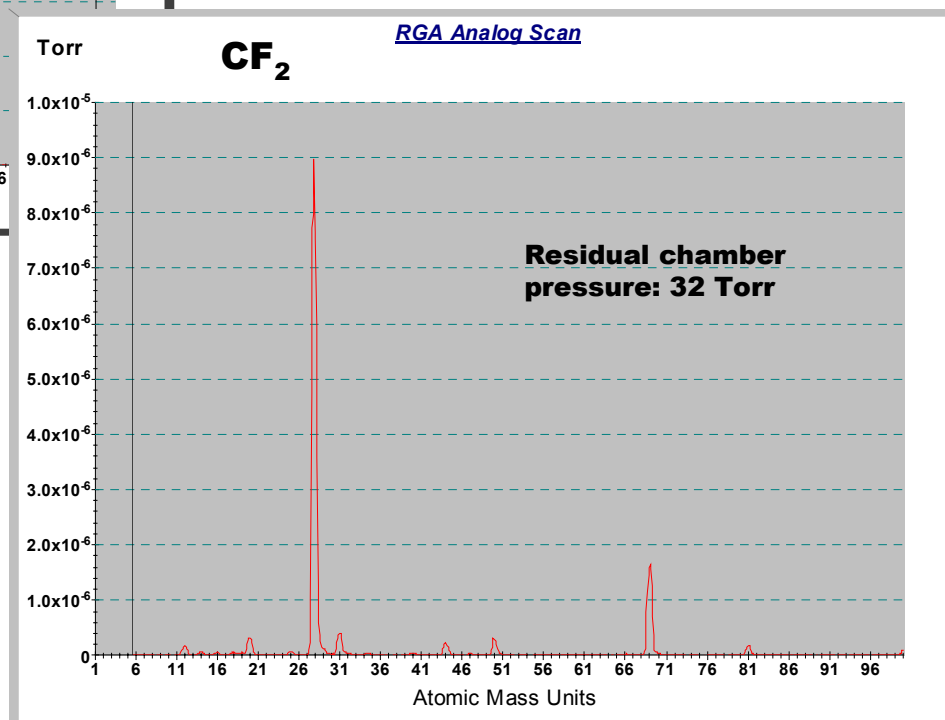
CO , CO_2 , Hydrocarbons: 1100 mTorr

N_2 : 180 mTorr

HO_2 : 40 mTorr

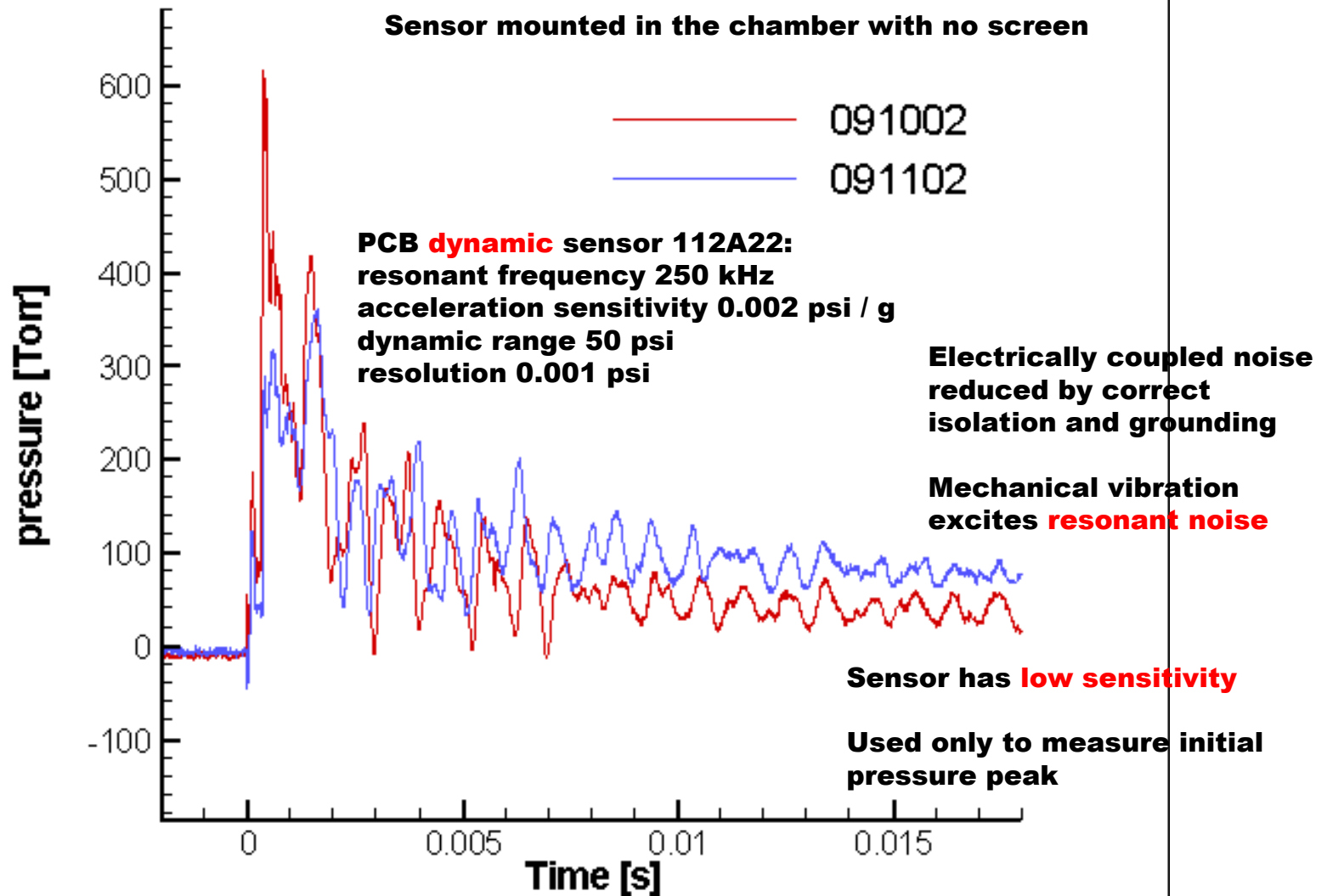
O_2 : 30 mTorr

LiF: 180 mTorr

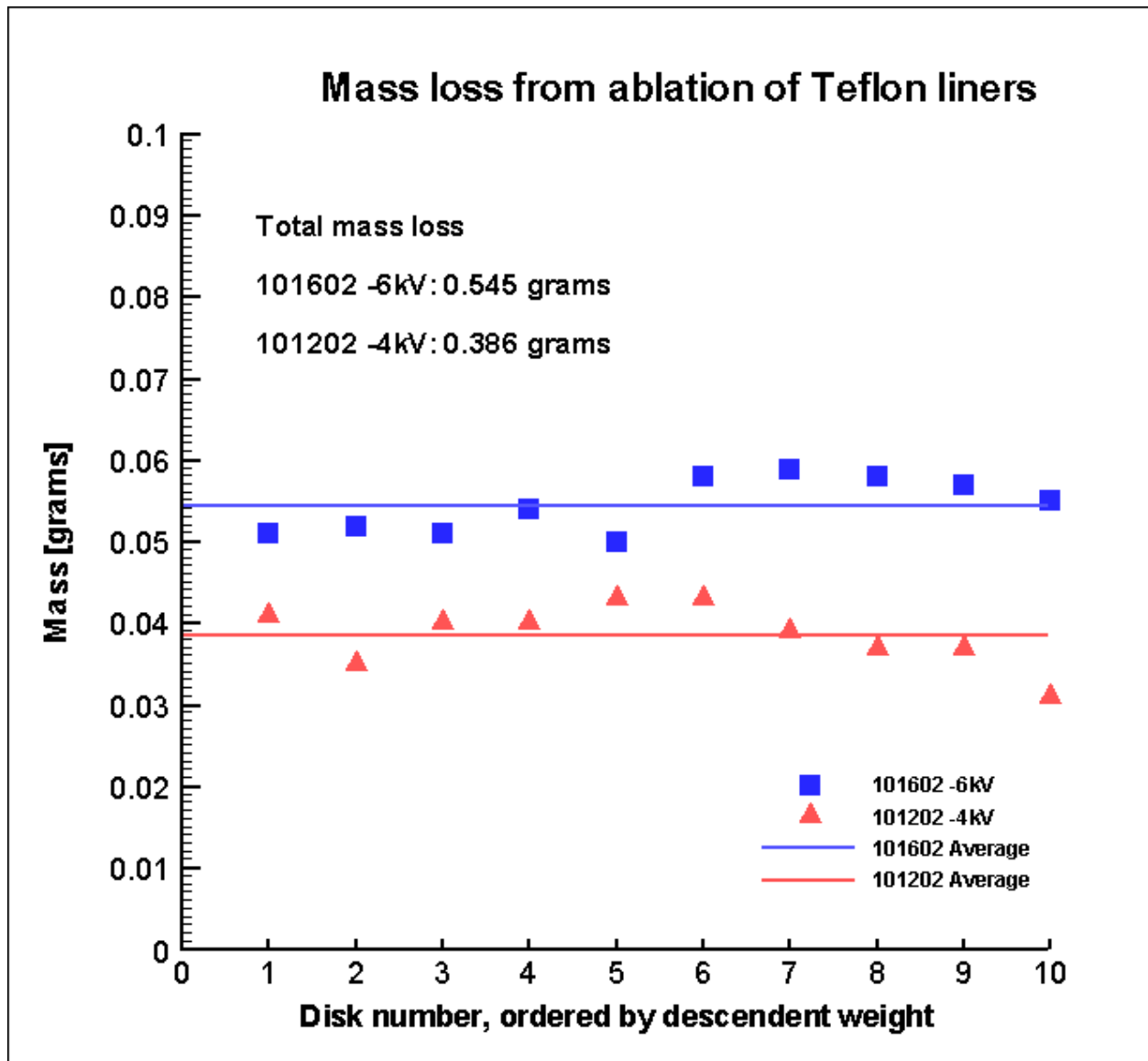


Chamber pressure history

Piezo-electric pressure sensor data for two 6 kV Teflon shots



Sleeve ablated mass: CF₂ experiments



The electro-thermal source

Characterization of superheated vapor generation shows measured parameters similar to SIRENS

Typical vapor parameters in the source :

$$n = 10^{19} - 10^{20} \text{ \# / cm}^3$$
$$T = 1-3 \text{ eV}$$

Close Ignitron switch



Cu triggering wire vaporizes (10-100 nanos) forming initial plasma column



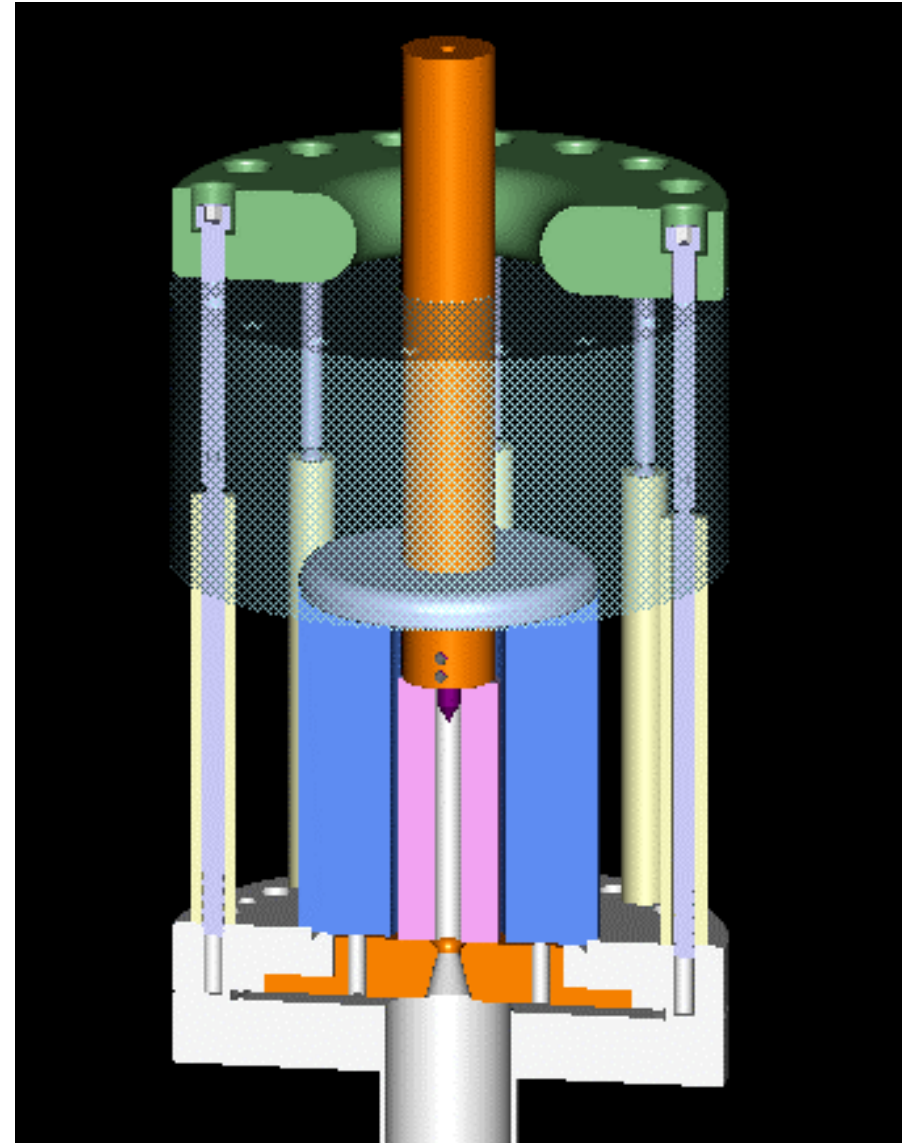
Energy stored in cap banks maintains plasma at 1-3 eV for 100 micros



Injected electrical power radiated to surface, ablates material of interest

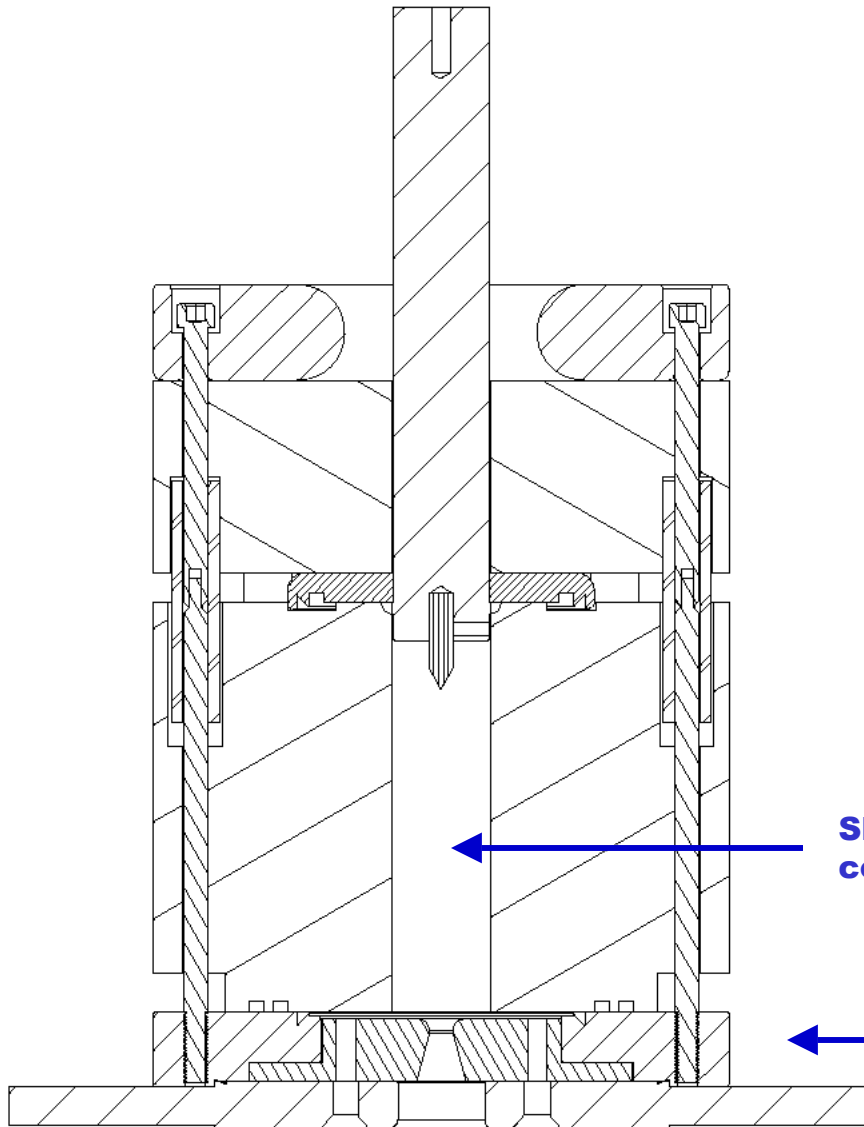
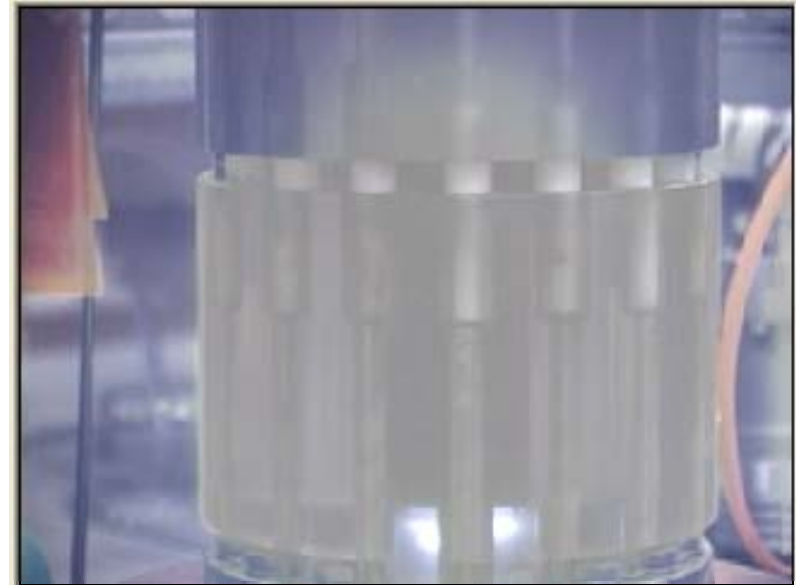


Pressure gradient drives injection, ablation balances axial mass loss



The electro-thermal source

LiF superheated vapor injection - low temperature materials configuration



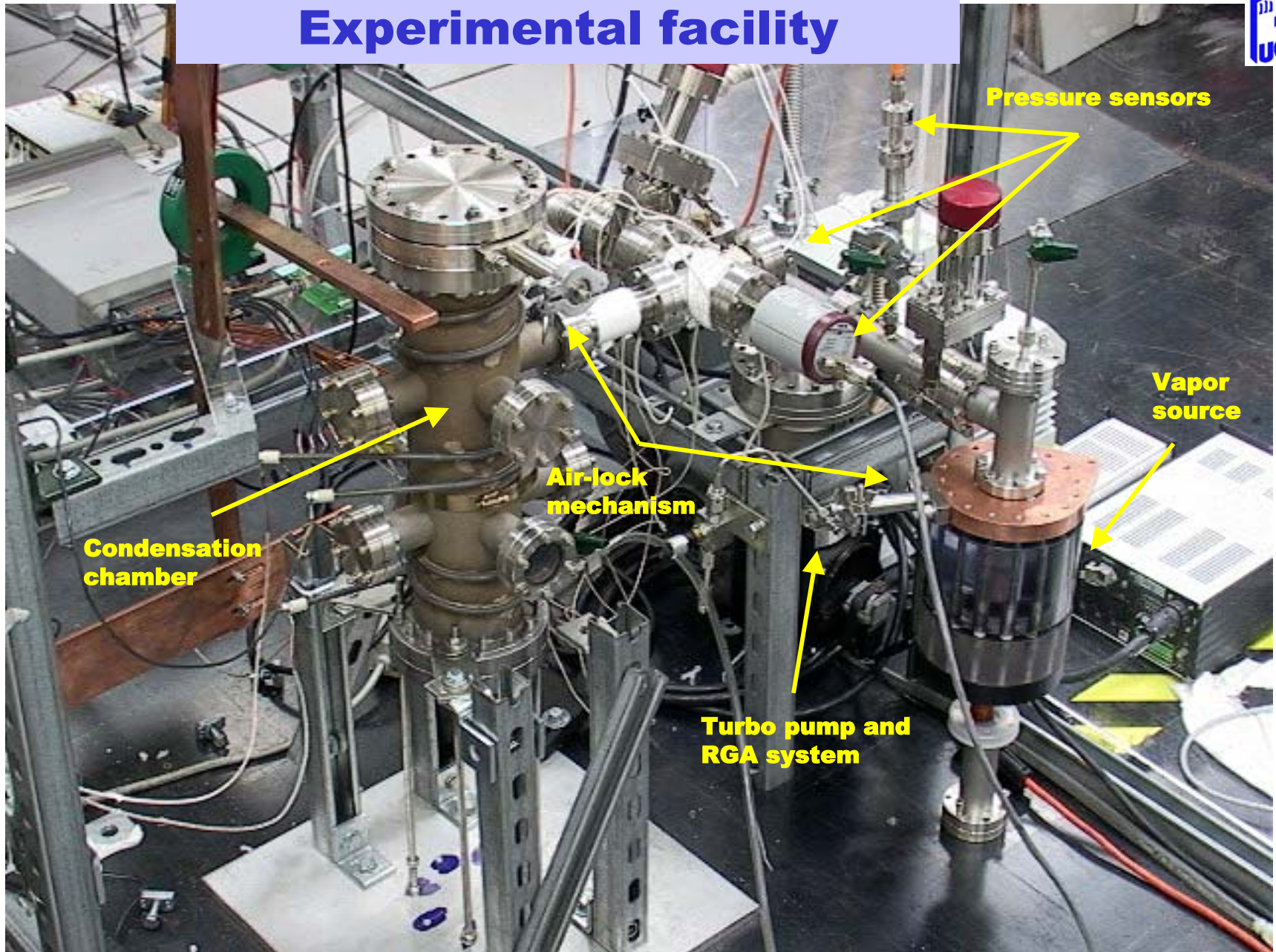
Sleeve (LiF) configuration

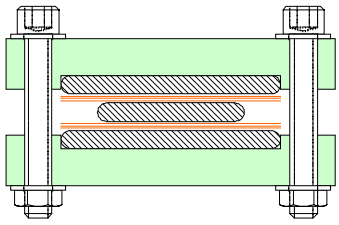


Chamber upper flange

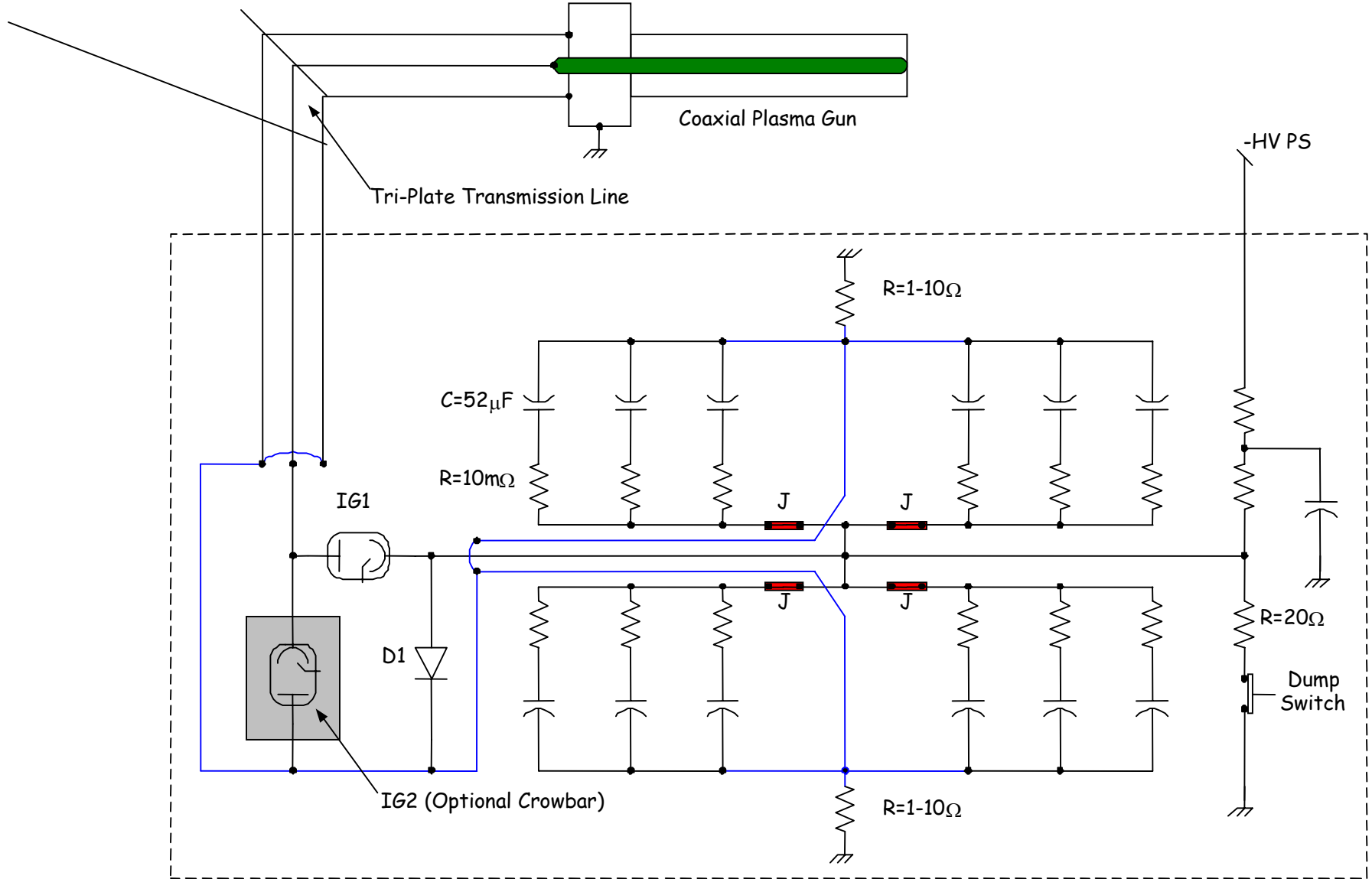
Connection to return leg and vacuum sealing

Experimental facility

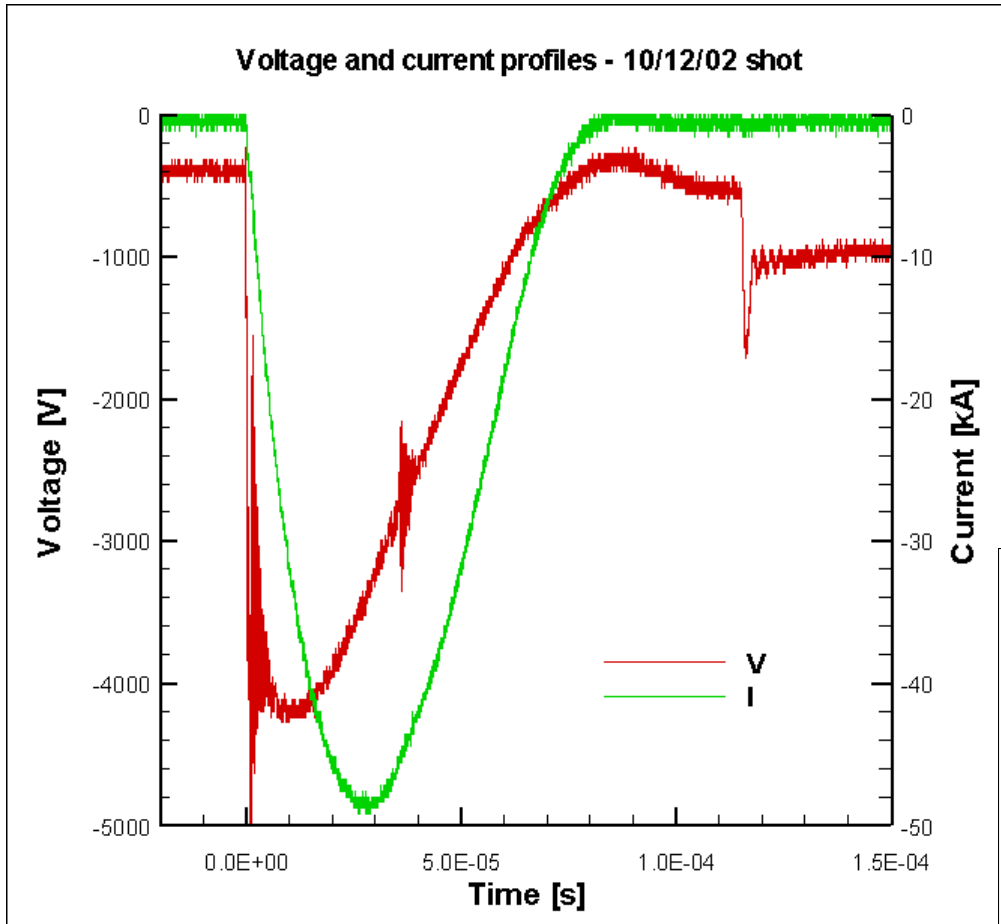




Pulse Forming Network scheme



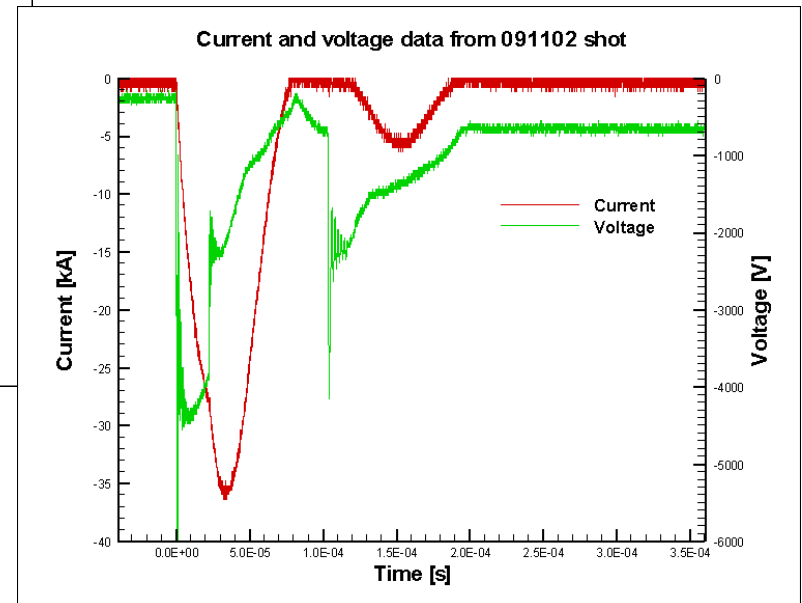
Measured discharge parameters



Triggering wire eliminated the possibility of secondary discharges

Enhanced reliability of the source is key for flibe

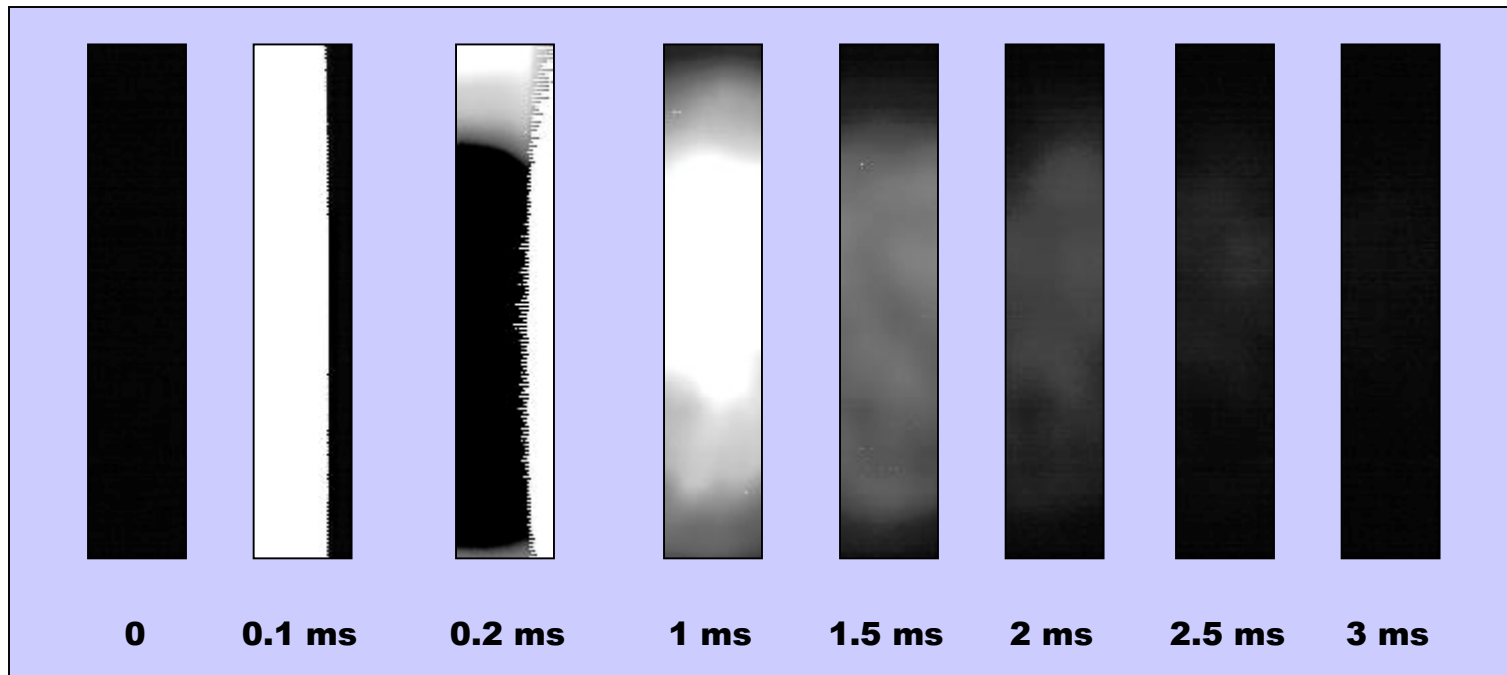
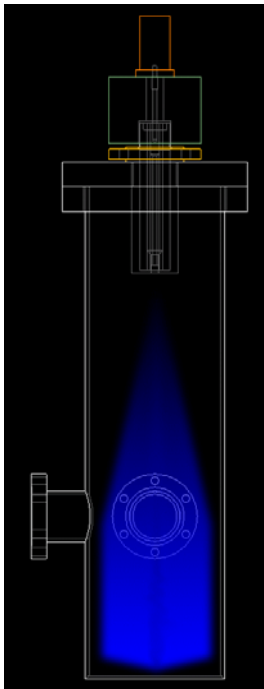
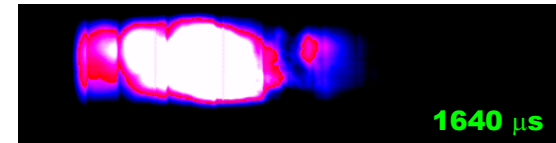
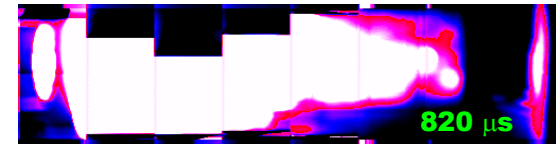
Measured electrical parameters of typical teflon shots during facility testing - power injected in the superheated vapor is $V \times I$



Light emission characterization

Lexan shots

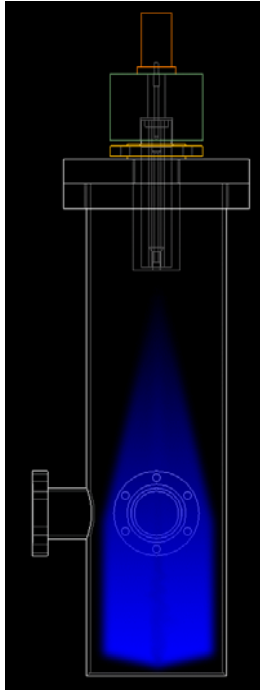
Frame sequences recorded with high speed camera - 10,000 frames per second and shutter speed of 100 μ s





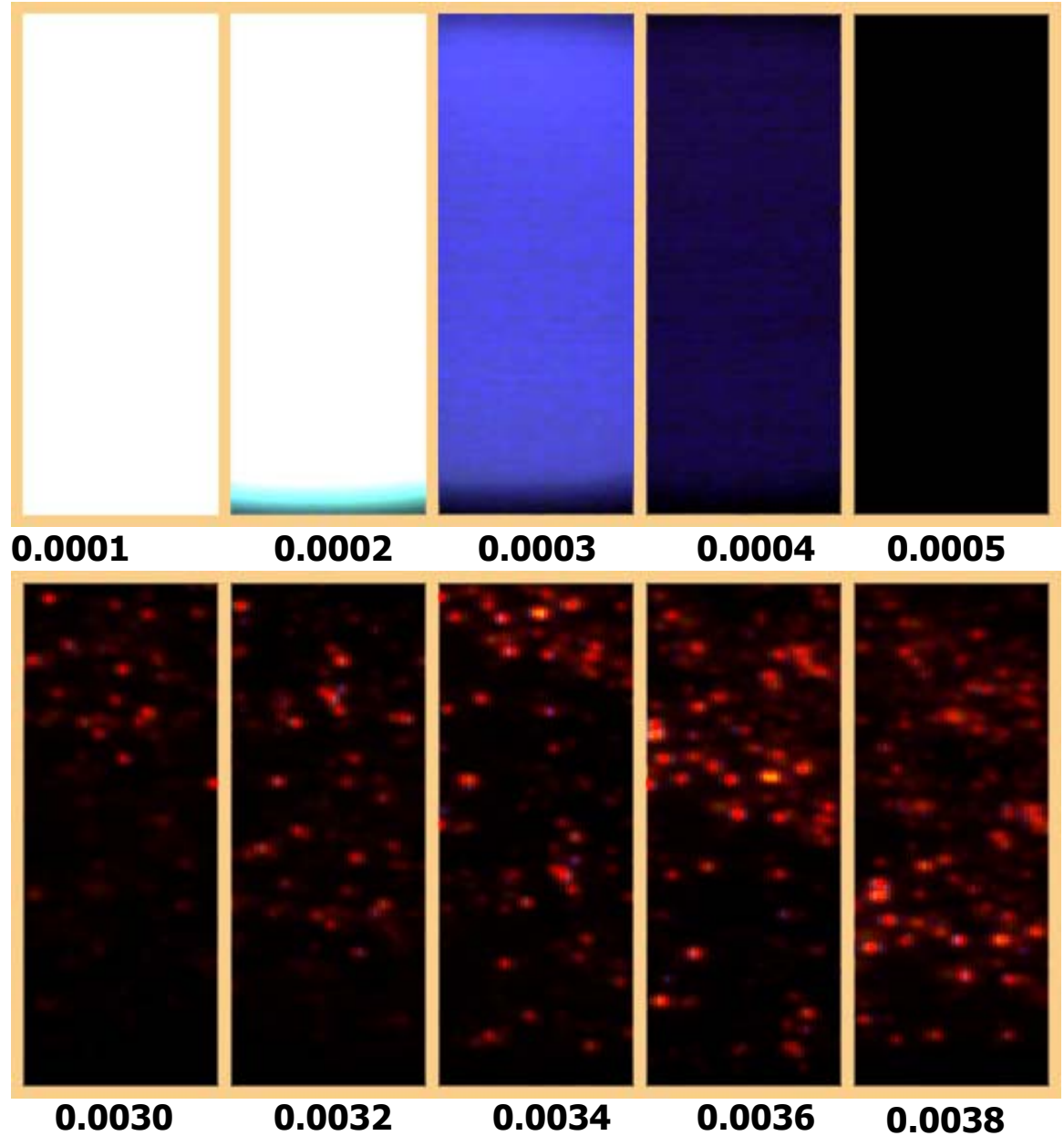
Light emission characterization

**Metal vapors characterization:
tin shot**



**With metal sleeves formation of a
liquid metal layer occurs over the
ablated surface**

**The liquid falls by gravity in the
chamber - droplet size too large to
be homogeneous volumetric
condensation**

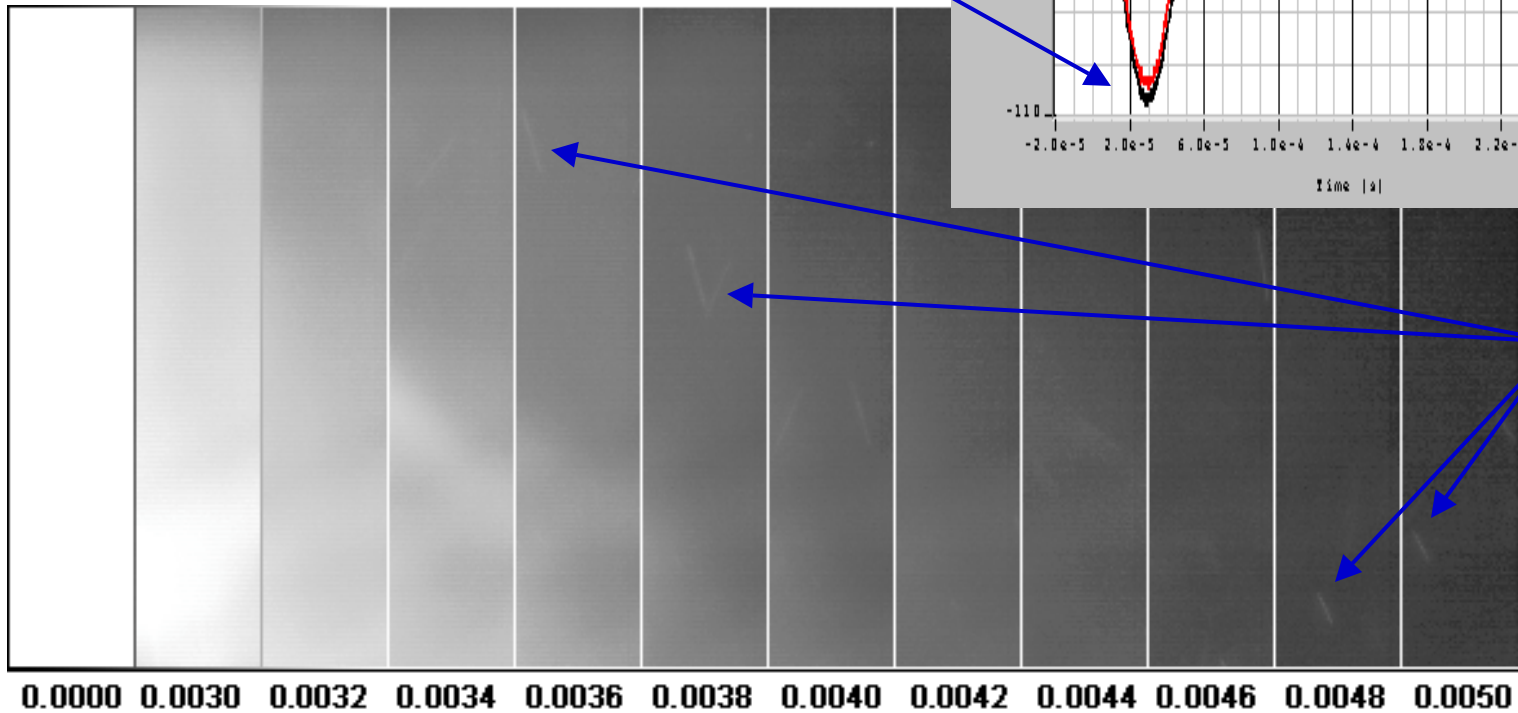
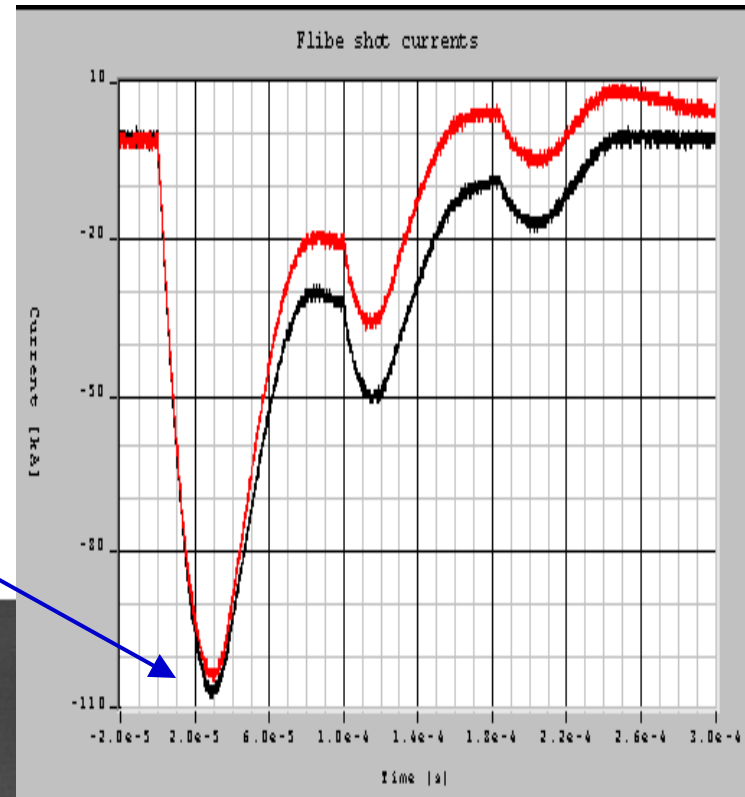


Light emission characterization

High energy flibe shot

Current peak 110 kA

Discharge completed 300 μ s after trigger



Injected debris

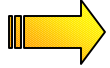
Emission spectroscopy



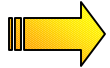
Objective:

Locally map the partial density of neutral Li, LiF and Be, BeF₂ as a function of time over the whole chamber clearing period

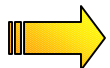
Neutral species temperature can also be inferred with additional development



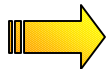
Injected vapor spontaneous emission is short lived, and superheated vapor properties are far from optimal for spectroscopy measurements - high density and low temperature



An high-voltage arc is used to re-excite the condensing vapor and analyze radiation emission at different times - excited vapor can reach up to 30 eV typically and densities will drop as the vapor condenses



The arc will excite the vapor locally, allowing to analyze the different properties of vapor in the volume bulk or near a condensing surface

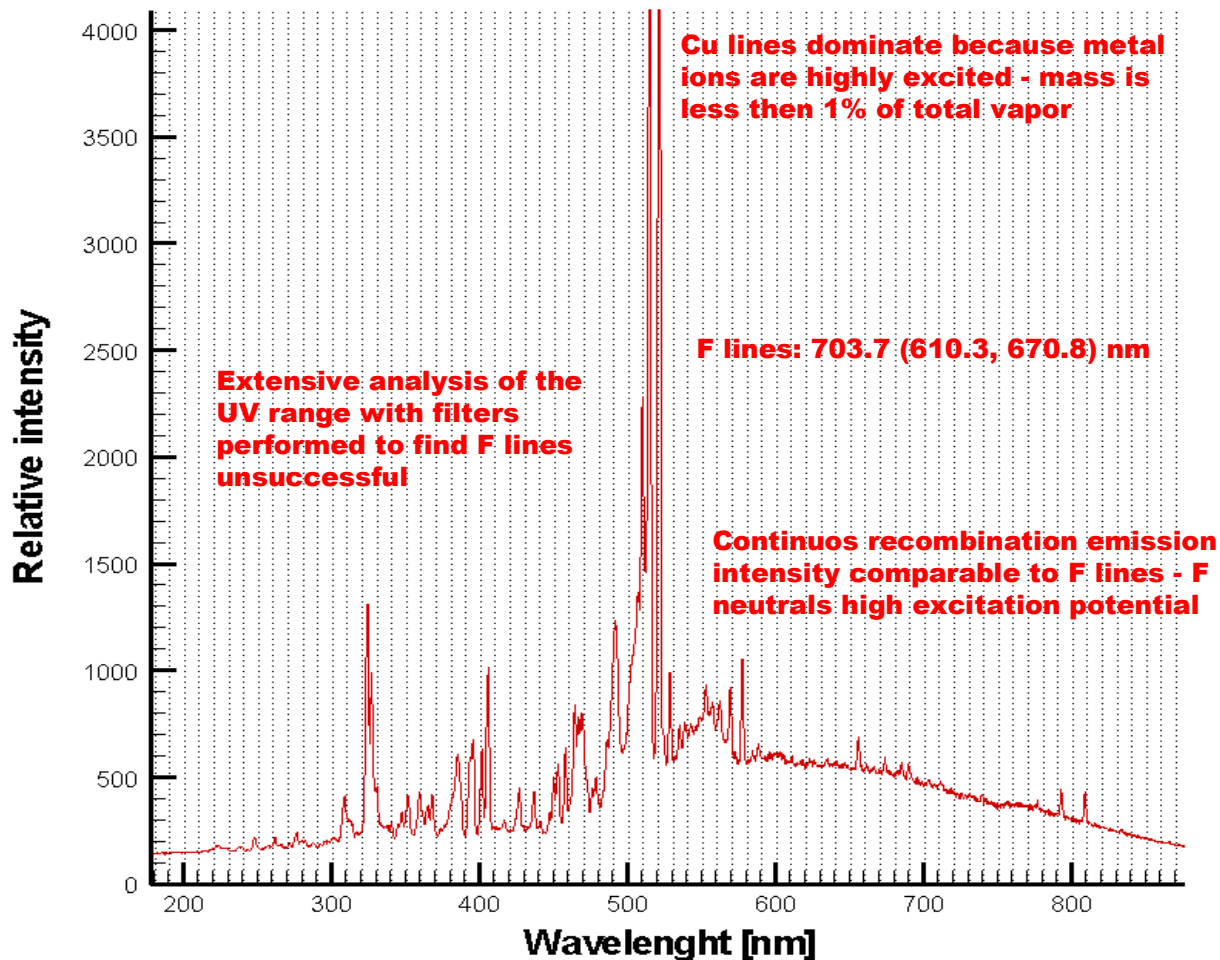


The high-voltage arc is under testing in steady state conditions with LiF vapors

Emission spectroscopy - CF₂ shot



Emission of CF₄ partially ionized gas integrated over 50 ms



Goal: find suitable lines for time resolved spectroscopy that can be analyzed with all materials of interest (F)

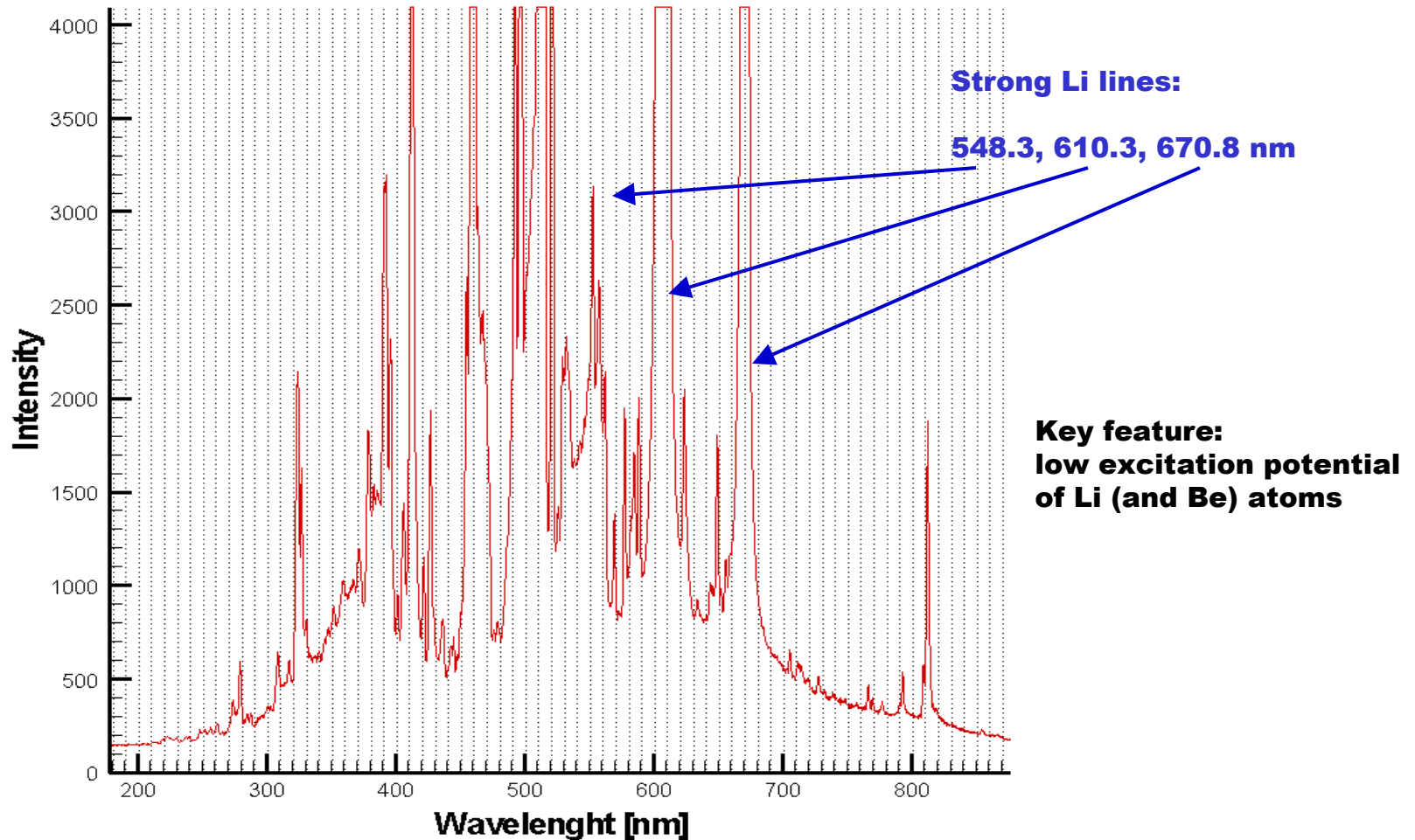
Due to non-uniform excited vapor generated with the triggering wire, Cu lines dominate the spectrum in the initial phase

Preliminary results of time resolved analysis of a Cu line with a photon multiplier showed metal emission is confined in the first millisecond

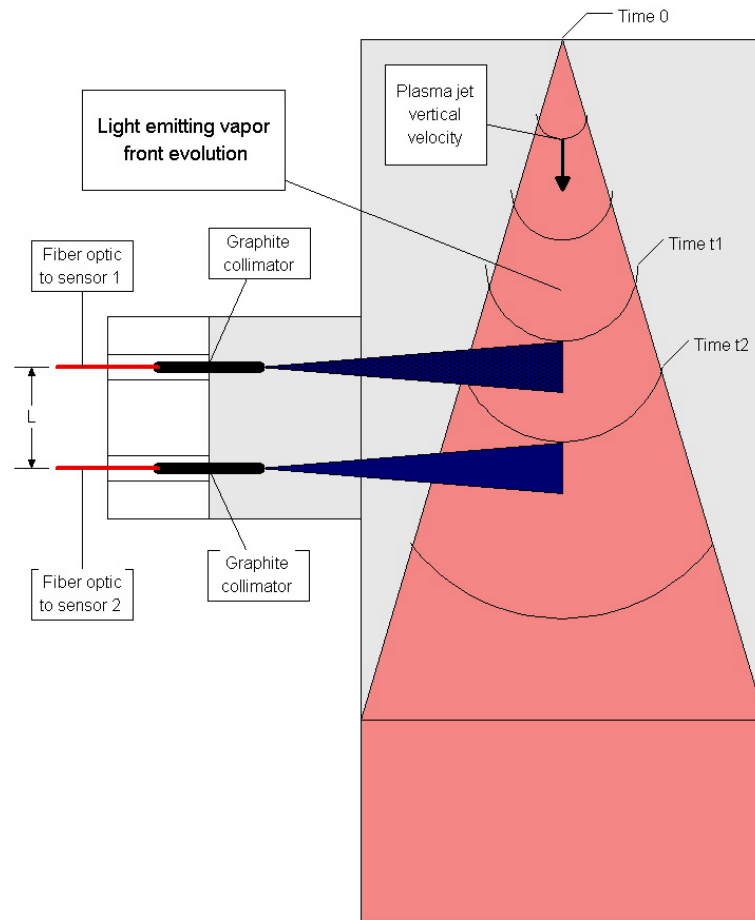
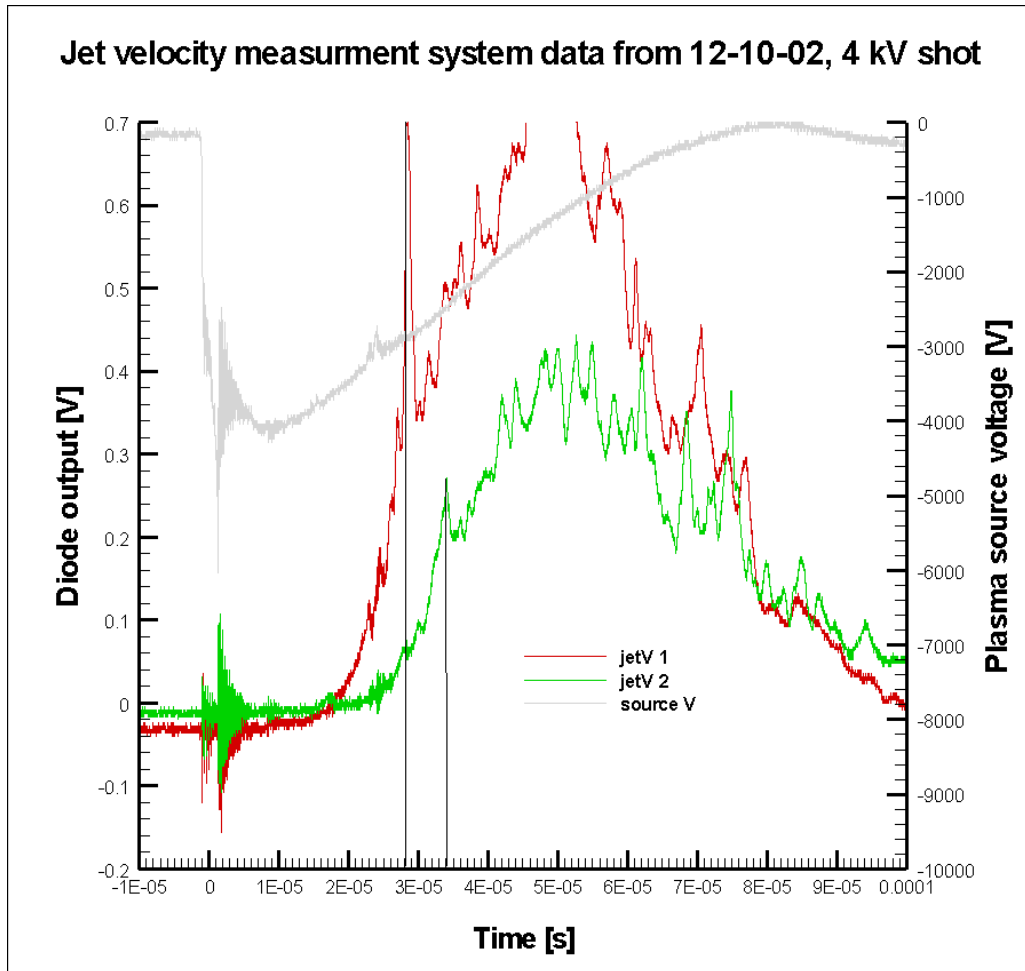
Emission spectroscopy - LiF shot



Emission from LiF partially ionized ions integrated over 50 ms



Jet velocity optical measurement system



Diode axis separation:

7.62 cm

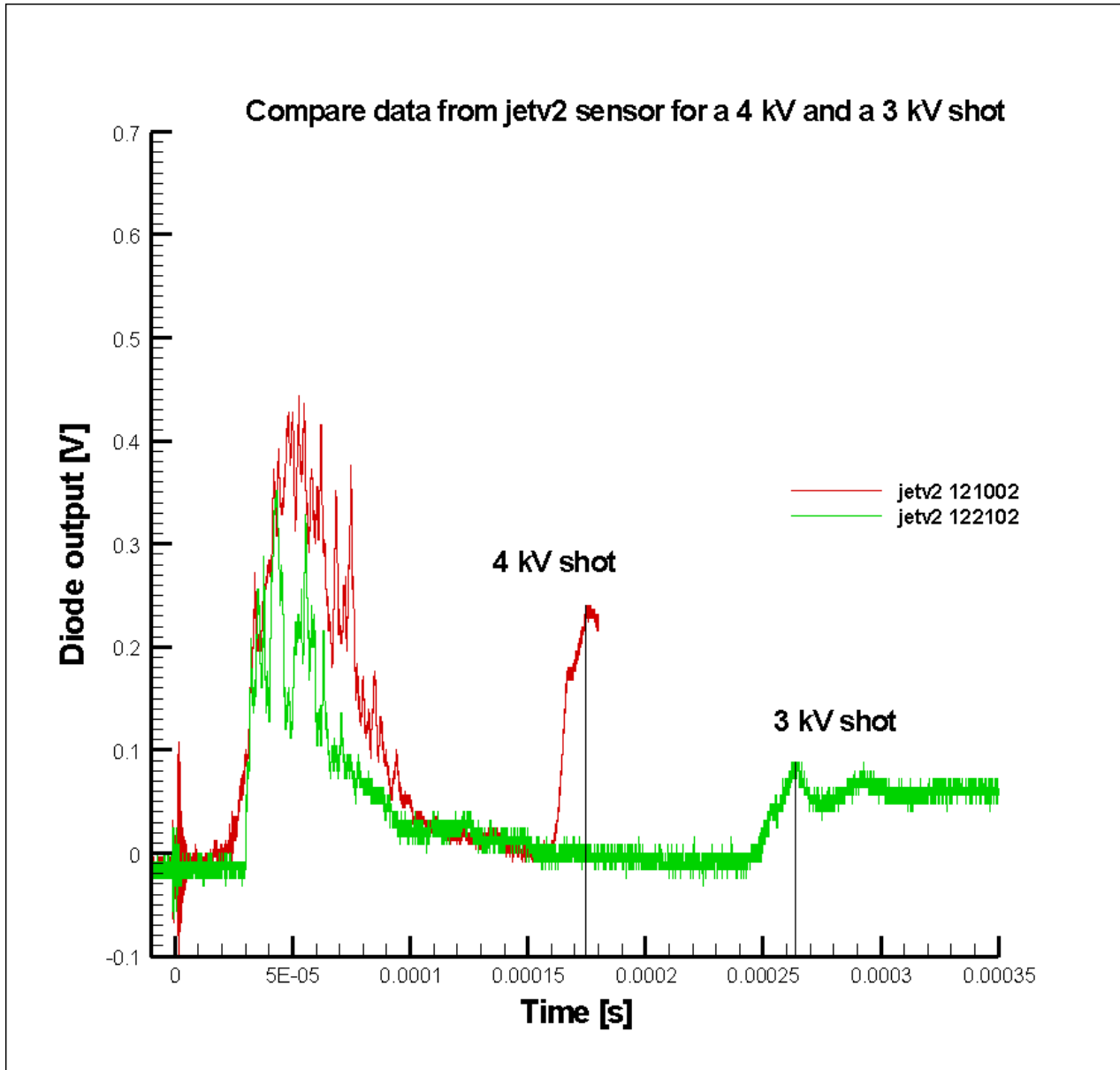
Peak time delay:

6 microseconds

Estimated initial vapor velocity:

12700 m/s

Jet velocity optical measurement system



Sensor closer to center of chamber sees a second peak in the light, about 2ms after triggering

Peak due to first reflection of the vapor jet at the chamber bottom

Vapor cools and expands in the chamber, emitting front does not reach the upper sensor

**Estimated average velocity of vapor in the chamber after first reflection:
320 m/s - 4 kV
210 m/s - 3 kV**

Experimental plan



Reduce to minimum traces of non condensable gases in the chamber - extend pressure history to mTorr range:

- ➡ **new, clean chamber for LiF shots - no C contamination**
- ➡ **add air-lock system in the plasma source to avoid open chamber to air each shot to change triggering wire**



Measure LiF chamber clearing rates and characterize amount and composition of non condensable gas to define accurate simulations conditions for Tsunami / condensation code

Complete development of emission spectroscopy system - measure density of Li atoms and LiF molecules at different times in the bulk volume and near a condensing surface - analyze data and provide conditions for accurate simulations with Tsunami / condensation code



Insert flibe disks in the source - measure the effect of the added presence of BeF₂ in the vapor

The End